

## 020508 Quiz 6 Nanoparticles

1)(35pts)

- a) **Show** that the terminal sedimentation velocity,  $c_s$ , is proportional to  $d_p^2$  in the continuum regime **and** to  $d_p$  in the free molecular regime.
- b) The Langevin equation is used to describe the particle flux,  $J_x$ , in terms of the concentration gradient and particle velocity.

**Write the 1-d Langevin equation for sedimentation.**

- c) **Plot** log sedimentation flux versus log particle size for the diffusion and sedimentation limits **indicating** the size for the minimum value **and** the transport regime you used. (**show** the slopes)
- d) For a sample with a wide unimodal particle size distribution from 10nm to 10 micron, **sketch** the particle size distribution for a sample collected from a sedimentation layer **and** the particle size distribution remaining in an aerosol or suspension.

2)(20pts)

- a) For an aerosol of particles smaller than 1 micron with a broad particle size distribution, **will** a representative sample be obtained by thermophoretic sampling?
- b) **Why?**

3)(45pts)

- a) **Write** the Smoluchowski equation.
- b) **Define** the terms.
- c) For particles undergoing Brownian motion, **give** the collision frequency function in terms of the diffusion coefficient and the particle size.
- d) **Give** the function for particles in the continuum range.
- e) **Give** the function for particles in the free molecular range.

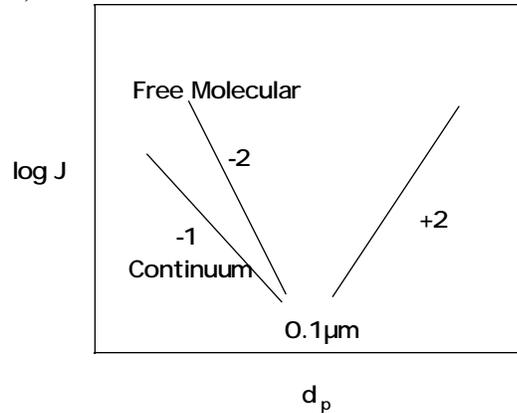
### Answers: 020508 Quiz 6 Nanoparticles

1) a)  $F_{\text{gravity}} = f c_s d_p^3 g$   
 so  $c_s = d_p^3 / f$

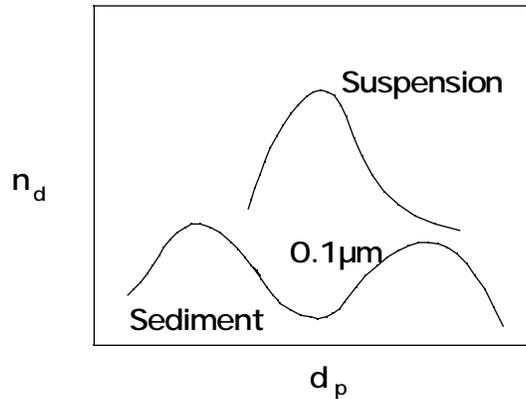
For the free molecular regime  $f \propto d_p^{-2}$  and for the continuum regime  $f \propto d_p$  following Stokes law.  
 Then for the continuum regime  $c_s \propto d_p^2$ , and for the free molecular regime  $c_s \propto d_p$

b)  $J_z = -D \frac{dn}{dz} - c_s n$

c)



d)



2) a) Yes

b) The thermophoretic deposition (velocity) is not particle size dependent in the free molecular range,  $c_T = (-3v_T) / (4(1 + \dots/8))$ .

3) a)  $\frac{dn_k}{dt} = \frac{1}{2} \sum_{i+j=k} (v_i, v_j) n_i n_j - n_k \sum_{i=1} (v_i, v_k) n_i$

b)  $n$  is the number concentration of particles,  $t$  is time,  $\sum$  is the collision frequency function,  $v$  is the particle volume.

c)  $\sum_{ij} (v_i, v_j) = 4 (D_i + D_j)(a_i + a_j)$

d)  $\sum_{ij} (v_i, v_j) = \frac{2kT}{3\mu} \left( \frac{1}{v_i^{1/3}} + \frac{1}{v_j^{1/3}} \right) (v_i^{1/3} + v_j^{1/3})$  **Continuum**

e)  $\sum_{ij} (v_i, v_j) = \frac{3}{4} \frac{6kT}{p} \left( \frac{1}{v_i^{1/2}} + \frac{1}{v_j^{1/2}} \right) (v_i^{1/3} + v_j^{1/3})^2$  **Free Molecule**